



RAN-2103000205021003

T.Y.B.Sc. (Sem.-V) Examination October - 2025

Physics : Paper-VIII (old Course) Atomic and Nuclear Physics

Time: 2 Hours]
[Total Marks: 50
સૂચના : / Instructions

- (1) ઉપરોક્ત દર્શાવેલ વિગતો ઉત્તરવહી પર અવશ્ય લખવી.
Fill up strictly the above details on your answer book

Seat No.:

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Student's Signature

Q.1. Answer in short as directed:
10

- Name three quantum numbers that arises in solution of Schrodinger's equation for hydrogen atom.
- What are the possible values of magnetic quantum number m_l of an electron in p-state?
- Which quantum number figures orientation of electron orbit?
- State condition of allowed transition for an electron in terms of wave function.
- What is Bohr magneton?
- What is meant by anti-symmetric wave functions?
- What are magic numbers?
- What is electron capture?
- What is meant by disintegration energy of nucleus?
- A nucleus ${}^A_Z X$ disintegrate to ${}^{A'}_{Z'} Y$ through an α decay followed by a β^- decay then $Z' = \text{-----}$ and $A' = \text{-----}$

Q.2. (a) Attempt any one of the following:
07

- From Schrodinger's equation of hydrogen atom in spherical polar coordinates, obtain three separate equations for each variable r , θ , and ϕ
- Discuss radial part of the Schrodinger's equation and obtain formula for electron's angular momentum $L = \sqrt{l(l+1)} \hbar$.

(b) Attempt any one of the following:
03

- The wave function of 1s electron of hydrogen atom is given by $\Psi(r, \theta, \phi) = \frac{e^{-r/a}}{\sqrt{\pi a^3}}$. Verify that $\left\langle \frac{1}{r} \right\rangle = \frac{1}{a}$.
- Discuss uncertainty principal and space quantization of electron in hydrogen atom.

Q.3. (a) Attempt any one of the following:
07

- What is Zeeman Effect? Explain normal Zeeman Effect with the help of vector atom model.
- Explain radiative transitions of hydrogen atom with the help of wave function and justify Bohr's assumption about it.

- (b) Attempt any one of the following:** **03**
- (1) A beam of electrons enters a uniform magnetic field of 1.20T. Find the energy difference between electrons whose spins are parallel and antiparallel to the field. (Take $\mu_B = 5.79 \times 10^{-5} \text{ eV/T}$ and $h = 6.625 \times 10^{-34} \text{ J-s}$)
 - (2) Discuss selection rules for allowed transitions.
- Q.4. (a) Attempt any one of the following:** **07**
- (1) Derive the expression for the Weizsacker's semi-empirical mass formula of the nucleus
 - (2) Explain in detail about the volume energy term (B_1) and surface energy term (B_2) about the nucleus.
- (b) Attempt any one of the following:** **03**
- (1) Calculate the total binding energy of ${}^{40}_{20}\text{Ca}$ in Mev. (Mass of the ${}^{40}_{20}\text{Ca}$ is 39.962591 amu, mass of proton and neutron are 1.007825 amu and 1.008665 amu respectively)
 - (2) Find the value of Coulomb energy term for ${}^{28}_{14}\text{Si}$. Proportionality constant for Coulomb energy term is 0.7 MeV.
- Q.5. (a) Attempt any one of the following:** **07**
- (1) What is radioactive decay? Explain Gamow theory of alpha decay.
 - (2) Explain β -decay. Derive necessary condition for spontaneous emission of β^- particle.
- (b) Attempt any one of the following:** **03**
- (1) Write short note on Geiger-Nuttall law.
 - (2) The total energy liberated in the α -decay of ${}^{226}_{88}\text{Ra}$ is 4.87MeV. Find the kinetic energy of an α -particle and the recoil energy of the nucleus.